

# Combination ISE

Ion-selective electrodes are responsive to concentration of particular ions in the test liquid and are variable-potential electrodes. They are used in conjunction with reference electrodes to measure the concentration of particular ions. HORIBA's years of experience and know-how in this field are behind the wide range of ion electrodes we offer.

When measurements are made using an ion meter, calibrating it with various standard solutions will give direct readings of the ion concentration. Note that since volume-detection level changes with temperature, measurements must be taken at a fixed temperature.

Type	Measurement Range	Selection Coefficient
<p>Ammonia ion electrode (combination) <b>5002A-10C</b></p>  <p>3014093560</p> <p>Overall length: 161 mm Diameter of probe: 15 mm Connector: BNC</p>	<ol style="list-style-type: none"> <li>0.1 to 1,000 mg/L NH<sub>3</sub></li> <li>Adjust more than pH 12</li> <li>0 to 50°C</li> <li>Within 30 seconds when substituting low concentration to high concentration Within 2 minutes when substituting high concentration to low concentration</li> </ol>	<p>_____</p>
<p>Chloride ion electrode (combination) <b>6560-10C</b></p>  <p>3014093430</p> <p>Overall length: 150 mm Diameter of probe: 16 mm Connector: BNC</p>	<ol style="list-style-type: none"> <li>0.35 to 35,000 mg/L Cl<sup>-</sup> (10<sup>-5</sup> to 1 mol/L Cl<sup>-</sup>)</li> <li>350 mg/L (10<sup>-2</sup> mol/L) Cl<sup>-</sup>, pH 3 to 11</li> <li>0 to 50°C</li> <li>Within 5 seconds</li> </ol>	<p>S<sub>2</sub>O<sub>3</sub><sup>2-</sup>, S<sup>2-</sup>, I<sup>-</sup>, Ag<sup>+</sup>, Hg<sup>2+</sup> = Not acceptable SCN<sup>-</sup> = 0.3, MnO<sub>4</sub><sup>-</sup> = 0.1 Br<sup>-</sup> = 0.03 NO<sub>3</sub><sup>-</sup>, F<sup>-</sup>, HCO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, PO<sub>4</sub><sup>2-</sup> = 1,000</p>
<p>Fluoride ion electrode (combination) <b>6561-10C</b></p>  <p>3014093431</p> <p>Overall length: 150 mm Diameter of probe: 16 mm Connector: BNC</p>	<ol style="list-style-type: none"> <li>0.02 to 19,000 mg/L F<sup>-</sup> (10<sup>-6</sup> to 1 mol/L F<sup>-</sup>)</li> <li>20 mg/L (10<sup>-3</sup> mol/L) F<sup>-</sup>, pH 4 to 10</li> <li>0 to 50°C</li> <li>Within 5 seconds</li> </ol>	<p>Possible interference when multiply-charged ion (ex. Al<sup>3+</sup>, Fe<sup>3+</sup>) coexisted and foamed the complex.</p>
<p>Nitrate ion electrode (combination) <b>6581-10C</b></p>  <p>3014093432</p> <p>Overall length: 150 mm Diameter of probe: 16 mm Connector: BNC</p>	<ol style="list-style-type: none"> <li>0.62 to 62,000 mg/L NO<sub>3</sub><sup>-</sup> (10<sup>-6</sup> to 1 mol/L NO<sub>3</sub><sup>-</sup>)</li> <li>62 mg/L (10<sup>-3</sup> mol/L) NO<sub>3</sub><sup>-</sup>, pH 3 to 7</li> <li>0 to 50°C</li> <li>Within 15 seconds</li> </ol>	<p>ClO<sub>4</sub><sup>-</sup> = 0.03, I<sup>-</sup> = 0.1, Br<sup>-</sup> = 2 NO<sub>2</sub><sup>-</sup> = 3, Cl<sup>-</sup> = 40, F<sup>-</sup> = 200 CH<sub>3</sub>COO<sup>-</sup> = 300, SO<sub>4</sub><sup>2-</sup> = over 1,000</p>
<p>Potassium ion electrode (combination) <b>6582-10C</b></p>  <p>3014093433</p> <p>Overall length: 150 mm Diameter of probe: 16 mm Connector: BNC</p>	<ol style="list-style-type: none"> <li>0.04 to 39,000 mg/L K<sup>+</sup> (10<sup>-6</sup> to 1 mol/L K<sup>+</sup>)</li> <li>3.9 mg/L (10<sup>-4</sup> mol/L) K<sup>+</sup>, pH 5 to 11</li> <li>0 to 50°C</li> <li>Within 15 seconds</li> </ol>	<p>Rb<sup>+</sup> = 0.4, Cs<sup>+</sup> = 3, NH<sub>4</sub><sup>+</sup> = 70 Li<sup>+</sup>, Na<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, Sr<sup>2+</sup>, Ba<sup>2+</sup> = over 1,000</p>
<p>Calcium ion electrode (combination) <b>6583-10C</b></p>  <p>3014093434</p> <p>Overall length: 150 mm Diameter of probe: 16 mm Connector: BNC</p>	<ol style="list-style-type: none"> <li>0.4 to 40,080 mg/L Ca<sup>2+</sup> (10<sup>-5</sup> to 1 mol/L Ca<sup>2+</sup>)</li> <li>4.0 mg/L (10<sup>-4</sup> mol/L) Ca<sup>2+</sup>, pH 5 to 11</li> <li>0 to 50°C</li> <li>Within 15 seconds</li> </ol>	<p>Fe<sup>3+</sup> = 0.1, Fe<sup>2+</sup>, Zn<sup>2+</sup> = 1, Sr<sup>2+</sup> = 50 Ni<sup>2+</sup>, Cu<sup>2+</sup> = 70, Co<sup>2+</sup> = 350 Mn<sup>2+</sup> = 500, Mg<sup>2+</sup> = 1,000 Na<sup>+</sup>, K<sup>+</sup>, Ba<sup>2+</sup>, NH<sub>4</sub><sup>+</sup> = over 1,000</p>